REACTION OF DIHYDROPYRAN WITH PHENYL AZIDE

R. A. Karakhanov, N. M. Shekhtman, and N. S. Zefirov

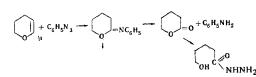
Khimiya Geterotsiklicheskikh Soedinenii, Vol. 5, No. 1, pp. 24-25, 1969

UDC 547.811'812.5

The reaction of dihydropyran with phenyl azide leads to the formation of N-phenyliminotetrahydropyran.

Reaction of aromatic azides with olefins takes place in two main directions depending on the strain of the double bond: with formation of triazolines in the case of strained olefins [1-4] and with the formation of arylimino derivatives for ordinary olefins [1, 2, 5]. For cyclic vinyl ethers with an unstrained double bond, this reaction should lead to aryliminolactones. The latter play an important part in processes of the cleavage of peptide bonds [6, 7].

We have found that the reaction of dihydropyran with phenyl azide does actually lead to an unstable, readily polymerizing, phenyliminolactone (I). The structure of I was shown by its hydrolytic decomposition.



The acid hydrolysis of I gives valerolactone, identified in the form of 5-hydroxyvaleric hydrazide. The alkaline saponification of I leads to the formation of aniline.

EXPERIMENTAL

Valerolactone phenylimide (I). A mixture of 4.25 g of phenyl azide and 10 g of dihydropyran was boiled for 10 hr. Vacuum distillation yielded 3.2 g of I with bp 135-140° C (3 mm). An analytical sample was obtained after two distillations, bp 130-132° C (2.5 mm), n_D^{20} 1.5707, d_4^{20} 1.100. MR_D 52.31. C₁₁H₁₃NO, Calculated: MR_D 51.97. Found, %: C 75.30; 75.28; H 7.67; 7.30; N 7.73; 7.89. Calculated for C₁₁H₁₃NO,%: C 75.40; H 7.48; N 7.99. The IR spectrum of I has an absorption band at 1667 cm⁻¹ (C=N) and lacks absorption bands in the 2170-2120 cm⁻¹ region (azide group).

A solution of 1 g of I in 7 ml of methylene chloride was shaken with 5 g of 25% sulfuric acid, the organic layer was separated off, the acid layer was extracted with 3 ml of methylene chloride, and the combined extracts were distilled in vacuum to give 0.21 g of δ -valerolactone with bp 112° C (13 mm), n_D^{20} 1.4691, d_4^{20} 1.079. According to the literature: bp 88°C (4 mm), mp 12.5° C, d_4^{20} 1.1081, n_D^{20} 1.4568 [8]; bp 83-85° C (3 mm), d_4^{20} 1.1439, n_D^{20} 1.4869 [9]. A 0.2 g quantity of δ -valerolactone was treated with

A 0.2 g quantity of δ -valerolactone was treated with 0.2 g of hydrazine hydrate. The crystals that deposited were pressed out on the filter, and recrystallization from a mixture of acetonitrile and benzene (1:1) gave 5-hydroxyvaleric hydrazide with mp 103-105° C. According to the literature: mp 107° C [10].

A mixture of 0.8 g of I and 5 g of 30% KOH solution was boiled for 18 hr and extracted with ether, and 1 ml of acetic anhydride was added to the ethereal extracts, which gave 0.32 g of acetanilide.

REFERENCES

1. K. Alder and G. Stein, Lieb. Ann., 501, 1, 1933.

2. K. Alder and G. Stein, Lieb. Ann., **515**, 185, 1934.

3. K. Alder, W. Günzl, and K. Wolff, Ber., 93, 809, 1960.

4. N. S. Zefirov, N. N. Kadzyauskas, and Yu. K. Yur'ev, ZhOKh, 35, 259, 1965.

5. G. Huisgen, L. Möbius, and G. Szeimies, Ber., 98, 1138, 1965.

6. B. Witkop, Advances of Protein Chemistry, 16, 221, 1961.

7. B. Cunningham and G. Schmir, J. Am. Chem. Soc., 88, 551, 1966.

8. R. Linstead and H. Rydon, J. Chem. Soc., 580, 1933.

9. C. Coffman, J. Am. Chem. Soc., 57, 1981, 1935.

10. W. Reppe, Lieb. Ann., 596, 125, 158, 1955.

9 December 1966 Zelinskii Institute of Organic Chemistry AS USSR

Moscow State University